The Use of Silyl Groups in Protecting the Hydroxyl Functions of Ribonucleosides

Kelvin K. Ogilvie, Krishan L. Sadana, Elaine A. Thompson,

Michael A. Quilliam, and John B. Westmore,

Department of Chemistry,

University of Manitoba,

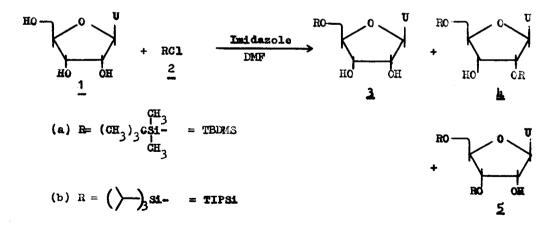
WINNIPEG, Manitoba, R3T 2N2,

Canada.

(Received in USA 22 May 1974; received in UK for publication 1 July 1974)

One of the most serious problems in the chemical synthesis of oligoribonucleotides involves the preparation of nucleosides bearing suitable protecting groups on the 2'-hydroxyl. Reese (1) and Smrt (2) have outlined in detail the problems of protecting groups in the ribonucleotide area. We wish to describe in this report the use of alkylsilyl groups for the protection of 2'and 2',5'-positions in ribonucleosides. These derivatives are readily prepared, are nicely crystalline, and overcome most of the problems associated with currently used protecting groups.

The two reagents that we have initially found to be most useful are the <u>tert</u>-butyldimethylsilyl (TBDMS) group and the tri-isopropylsilyl (TIPSi) group.



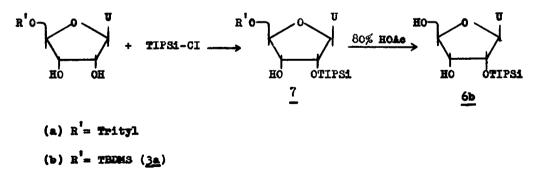
All reactions are carried out in DMF (1 ml/mmole of nucleoside) using two equivalents of imidazole per mmole of silyl chloride. Uridine reacts under these conditions with 1.2 equivalents of $\underline{2a}$ to produce $\underline{3a}$ (mp 136-139°C) in 73% yield. $\underline{4a}$ (mp 121-123°C) and $\underline{5a}$ (mp 137-139°C) are obtained in 11% and 7% yields respectively. Increasing the ratio of $\underline{2a:1}$ to 2.5:1 gives a 46% yield of $\underline{4a}$ and 32% yield of $\underline{5a}$. The tri-isopropylailyl chloride ($\underline{2b}$) is more selective. Thus using $\underline{2b:1}$ in the ratio of 1.2:1 gives a 76% yield of $\underline{3b}$ (mp 126-127°C) while a ratio of 2.5:1 produces $\underline{4b}$ (mp 146-147°C) and $\underline{5b}$ (mp 121-122°C) in 61% and 20% yields respectively. The isomeric compounds $\underline{4}$ and $\underline{5}$ separate cleanly on silica gel thick layer plates using ether or ether-hexane (1:1) as solvent.

We have also noted a large difference in rates of acid hydrolysis between a 2'(or 3') substituent and a 5'-substituent. This leads to a rapid and convenient preparation of 2'-monosilylated nucleosides. Thus heating <u>4a</u> with 80% HOAc for 20 min. on a steam bath gives 6a



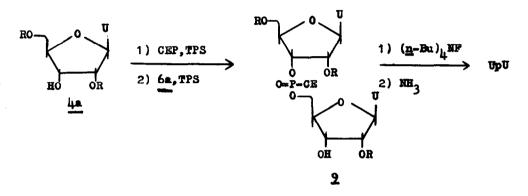
(mp 189-191°C) in 64% yield. We have not detected any isomerization of 4 and 5 under these acid conditions.

An excellent method for obtaining 2'-TIPSi-Uridine (6b) is shown below. When 5'-Trityl-



uridine is treated with 1.2 equivalents of TIPSi-Cl under the usual conditions <u>7a</u> (mp 125-126°C) is obtained in 70% yield. A 25% yield of the 3'-isomer (mp 80-82°C) is also obtained. Treatment of <u>7a</u> with 80% HOAc on a steam bath gives <u>6b</u> (mp 154-155°C) in 98% yield. If one starts with 5'-TBDMS-uridine under the same conditions a 70% yield of <u>7b</u> (mp 180-181°C) is obtained (12% of the 3'-isomer mp 106-107°C). Again acetic acid on a steam bath for 20 min quantitatively converts <u>7a</u> to <u>6b</u>. Thus compounds <u>6</u> and <u>7</u> are readily available and are clearly desirable derivatives for the present methods of nucleotide synthesis (1, 2).

We have synthesized a dinucleotide of uridine using the silyl protecting groups. Thus 4a was phosphorylated with β -cyanoethyl phosphate and condensed with <u>6a</u> in the usual manner (3).



Compound <u>9</u> was isolated on thick layer plates using ether as developing solvent. The silyl groups were removed with $(n-butyl)_4NF$ in THF while the cyanoethyl group was removed with ammonia (3). The resulting nucleotide uridylyl-(3'-5')-uridine (UpU) was completely degraded by both snake venom and spleen phosphodiesterases.

We will shortly describe the synthesis of important oligoribonucleotides sequences using only silyl groups for the protection of the hydroxyl functions.

Acknowledgement: We wish to acknowledge generous financial support from the National Research Council of Canada.

REFERENCES

- 1. D.P.L. Green, T. Ravindranathan, C.B. Reese and R. Saffhill, Tetrahedron, <u>26</u>, 1031 (1970).
- 2. J. Smrt, Coll. Czech. Chem. Comm., 38, 3189 (1973).
- 3. R. L. Letsinger and K.K. Ogilvie, J. Am. Chem. Soc., 89, 4801 (1967).